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Key indicators

Single-crystal X-ray study
T = 193 K
Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$
R factor = 0.038
wR factor = 0.103
Data-to-parameter ratio = 11.7

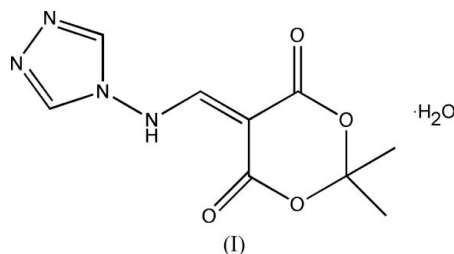
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

2,2-Dimethyl-5-(4*H*-1,2,4-triazol-4-ylamino-methylene)-1,3-dioxane-4,6-dione monohydrate

In the title compound, $\text{C}_9\text{H}_{10}\text{N}_4\text{O}_4 \cdot \text{H}_2\text{O}$, the triazole ring is nearly planar. The 1,3-dioxane-4,6-dione ring exhibits a half-chair conformation. Two intramolecular $\text{N}-\text{H} \cdots \text{O}$ hydrogen bonds, with $\text{O} \cdots \text{H}$ distances of 2.23 (2) and 1.99 (2) Å , and two intermolecular $\text{O}-\text{H} \cdots \text{N}$ hydrogen bonds, with $\text{N} \cdots \text{H}$ distances of 1.92 (3) and 1.94 (4) Å , are observed.

Comment

1,2,4-Triazole and its derivatives have been used as starting materials for the synthesis of many heterocycles (Desenko, 1995). Studies indicate that the 1,2,4-triazole group is associated with anti-inflammatory action (Gupta & Bhargava, 1978), and also with pharmacological activities, such as antiviral (Jones *et al.*, 1965), analgesic (Sughen & Yoloye, 1978), antimicrobial (Cansiz *et al.*, 2001), antidepressant (Kane *et al.*, 1988) and antifungal (Massa *et al.*, 1992). On the other hand, cyclic 1,3-diones like Meldrum's acid and their 5-arylamino-methylene analogs play an important role in heterocyclic chemistry as pivotal intermediates to access cyclic products (Gaber & McNab, 2001). We have already investigated Meldrum's acid derivatives as key intermediates to afford aza compounds with potential biological activity, such as phenanthrolines and pyrimidonaphthyridines (Silva *et al.*, 2002; Bortoluzzi *et al.*, 2005). As an extension of this approach, we report here an X-ray crystallographic study of the title compound, (I), within a project to investigate potential antiviral and Leish-manicidal activities.



In (I), the 1,3-dioxane-4,6-dione ring exhibits a half-chair conformation. The dihedral angle $\text{N}5-\text{N}6-\text{C}7-\text{C}8$ is $-177.2 (2)^\circ$ and the distances $\text{N}6-\text{C}7$ and $\text{C}7-\text{C}8$ indicate delocalization of the conjugated system. The molecular packing of (I) is stabilized by a hydrogen-bonded network (Fig. 2). Details of the hydrogen-bonding geometry are given in Table 1. The H atom of the NH group has one intramolecular contact to O13 and is also connected to the water O atom, indicating very weak $\text{N}-\text{H} \cdots \text{O}$ hydrogen bonding. Both H atoms of the water molecule are involved in $\text{O}-\text{H} \cdots \text{O}$ hydrogen bonding to two neighbouring molecules.

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Experimental

The title compound was prepared according to a literature procedure (Cassis *et al.*, 1985) and was recrystallized from acetone.

Crystal data

$C_9H_{10}N_4O_4 \cdot H_2O$
 $M_r = 256.23$
 Monoclinic, $P2_1/c$
 $a = 18.012$ (5) Å
 $b = 5.356$ (5) Å
 $c = 12.937$ (5) Å
 $\beta = 108.786$ (5)°
 $V = 1181.6$ (12) Å³
 $Z = 4$

$D_x = 1.440$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 25 reflections
 $\theta = 5.1$ – 20.4 °
 $\mu = 0.12$ mm⁻¹
 $T = 193$ (2) K
 Prism, colorless
 $0.50 \times 0.20 \times 0.10$ mm

Data collection

Enraf–Nonius CAD-4 diffractometer
 $\omega/2\theta$ scans
 Absorption correction: none
 2126 measured reflections
 2058 independent reflections
 1504 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.028$

$\theta_{max} = 25.1$ °
 $h = -21 \rightarrow 0$
 $k = 0 \rightarrow 6$
 $l = -14 \rightarrow 15$
 3 standard reflections
 every 200 reflections
 intensity decay: 1%

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.038$
 $wR(F^2) = 0.103$
 $S = 1.03$
 2058 reflections
 176 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0425P)^2 + 0.3902P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} < 0.001$
 $\Delta\rho_{max} = 0.25$ e Å⁻³
 $\Delta\rho_{min} = -0.15$ e Å⁻³
 Extinction correction: *SHELXL97*
 Extinction coefficient: 0.031 (2)

Table 1

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N6-H6 \cdots O13$	0.86 (2)	2.23 (2)	2.759 (2)	120 (2)
$N6-H6 \cdots O1W$	0.86 (2)	1.99 (2)	2.717 (3)	142 (2)
$O1W-H1WA \cdots N2^i$	0.91 (3)	1.92 (3)	2.822 (3)	174 (2)
$O1W-H1WB \cdots N3^{ii}$	0.96 (4)	1.95 (4)	2.872 (3)	160 (3)

Symmetry codes: (i) $x, -y + \frac{3}{2}, z - \frac{1}{2}$; (ii) $-x + 1, -y + 1, -z + 2$.

The amino and water H atoms were located in a difference map and were refined freely. H atoms on C atoms were positioned with idealized geometry and were refined with $U_{iso} = 1.2U_{eq}$ of the parent atom, or 1.5 for the methyl groups, using a riding model with $C-H = 0.93$ Å (0.96 Å for methyl groups).

Data collection: *CAD-4/PC Software* (Enraf–Nonius, 1993); cell refinement: *CAD-4/PC Software*; data reduction: *XCAD* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

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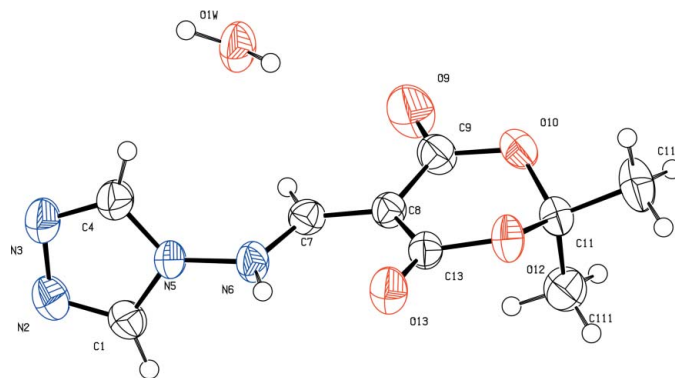


Figure 1

The molecular structure of (I), showing the atom labeling and displacement ellipsoids drawn at the 50% probability level.

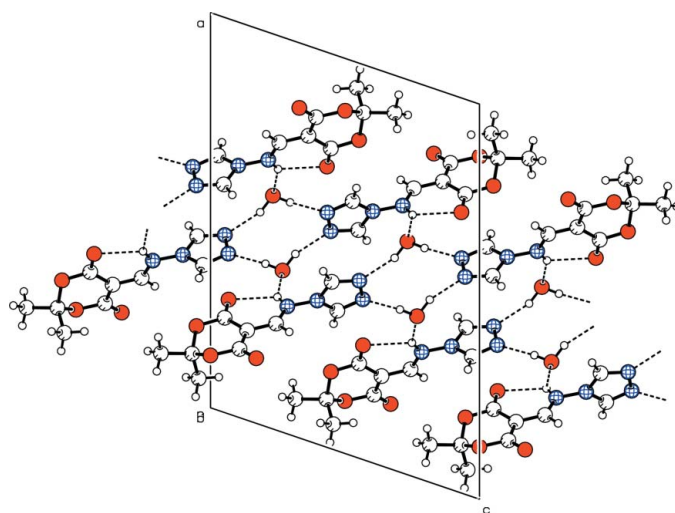


Figure 2

The molecular packing of (I), with hydrogen bonding shown as dashed lines.

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